

Daily Tutorial Sheet-15	Level-3
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159-161 159.(C) 160.(B) 161.(C)

 $A = KOH, B = H_2O_2, C = O_2, D = K_2SO_4, E = K_2CO_3$ 

162. [A-p, q, s] [B-p, q, r] [C-q, r] [D-q, r s]

 $\mathrm{CaH}_2\,$  reacts with water giving  $\,\mathrm{Ca}\left(\mathrm{OH}\right)_{\!2}\,$  and hydrogen.  $\mathrm{CaC}_2\,$  with water gives acetylene.

- **163.(3)** The alkali metal nitrate which gives  $NO_2$  gas on heating is  $LiNO_3$ . For Li, Z = 3.
- **164.(2)** In  $KHCO_3$ ,  $2HCO_3^-$  are joined by the hydrogen bonding.